

Apr. 11 2011

**Hokkaido University Special Lecture  
The Latest Information on the  
Fukushima Nuclear Power Plant**

**Environmental Radioactive Substances:  
Data from the School of Engineering**

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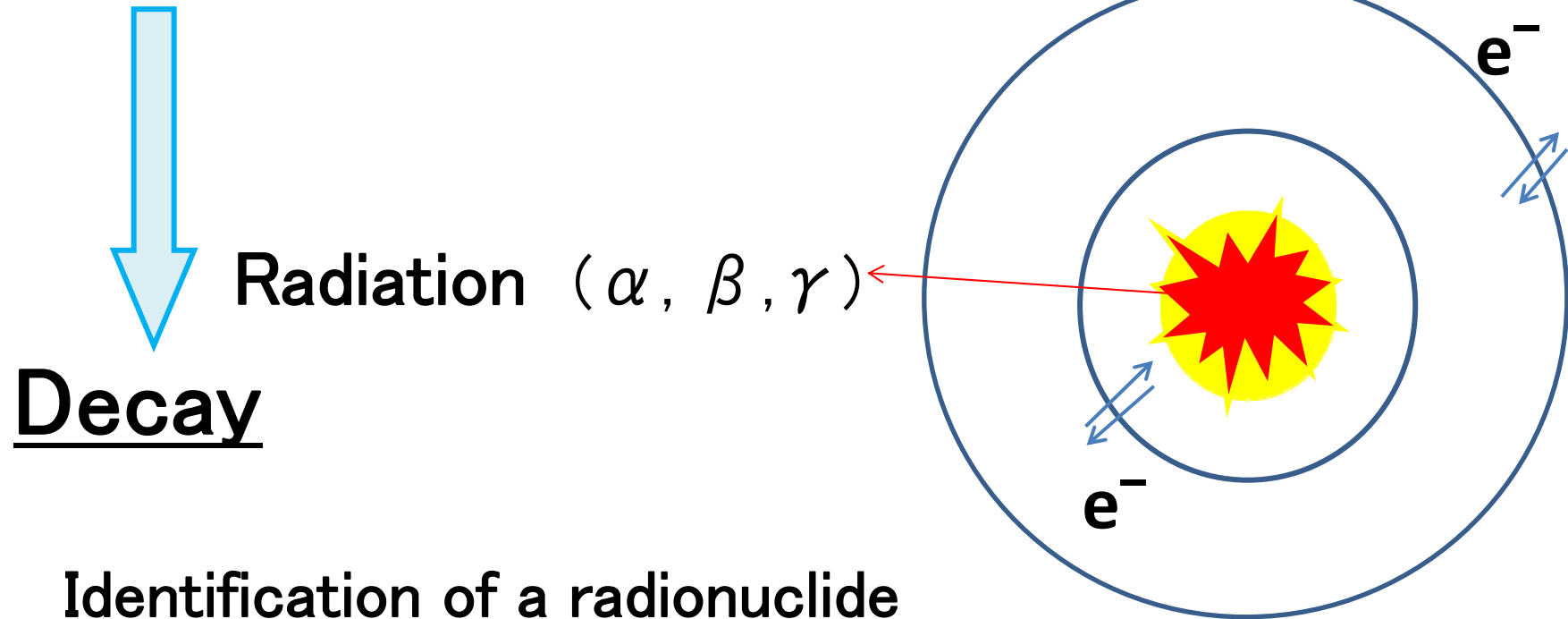
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# Contents

1. What is radioactive substances?
2. Radioactive substances in the environment?
3. What are measures of radioactive substance and radiation?
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5. Summary

# What is a radioactive atom (radionuclide)?

Atom with unstable (high energy level) nucleus



## Identification of a radionuclide

Emitting radiation	Type ( $\alpha, \beta, \gamma$ )
	Energy (E)
	Emission ratio
	Half life ( $T_{1/2}$ )

# What is radioactivity?

Number of radionuclides decaying in a unit time

Unit: **Becquerel (Bq)**

$$\frac{dN}{dt} = -\lambda N$$

Radioactivity (A) (Bq = s<sup>-1</sup>)

N: Number of radionuclides at time t

t: Time (s)

$\lambda$ : Decay constant

# How to represent a radionuclide

For example,

Radioactive potassium



(K-40, Potassium-40)

# There are many radionuclides in our environment!

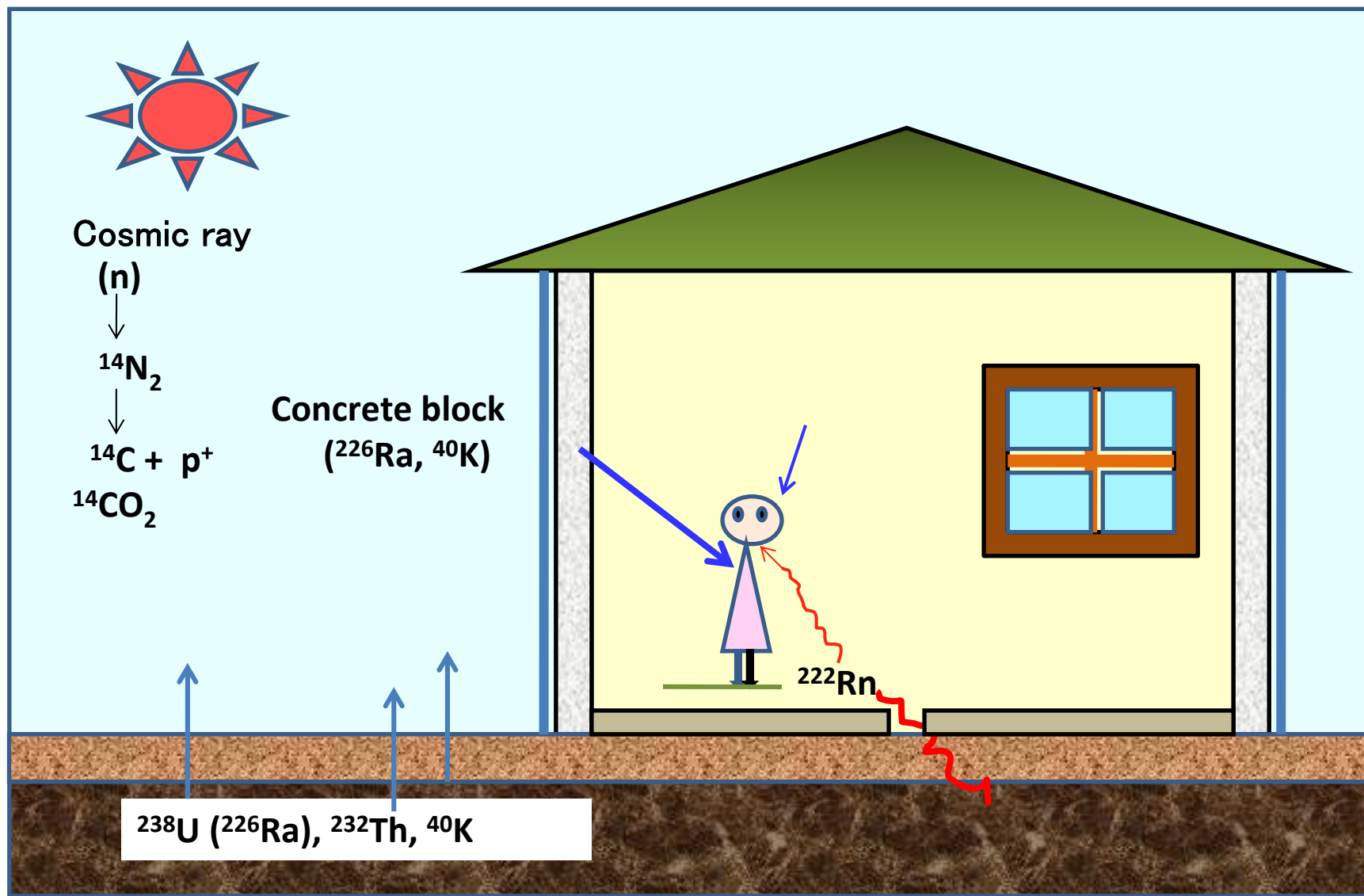
## Classification of radionuclides

1. Primordial radionuclides with long half life ( $T_{1/2} > 10^8$  y) ( $^{40}\text{K}$ ,  $^{238}\text{U}$ ,  $^{232}\text{Th}$ )
2. Cosmic ray produced radionuclides  
( $^3\text{H}$ ,  $^7\text{Be}$ ,  $^{14}\text{C}$  etc.)
3. Anthropogenic radionuclides  
( $^{60}\text{Co}$ ,  $^{90}\text{Sr}$ ,  $^{137}\text{Cs}$ ,  $^{131}\text{I}$ ,  $^{99\text{m}}\text{Tc}$  etc.)

# Natural Radionuclides in three Decay series ( $^{238}\text{U}$ , $^{232}\text{Th}$ , $^{235}\text{U}$ )

Element	U-238 Series							Th-232 Series				U-235 Series				
Neptunium																
Uranium	U-238 $4.47 \times 10^9$ yrs		U-234 $2.48 \times 10^5$ yrs									U-235 $7.04 \times 10^8$ yrs				
Protactinium		Pa-234 1.18 min											Pa-231 $3.25 \times 10^4$ yrs			
Thorium	Th-234 24.1 days		Th-230 $7.52 \times 10^4$ yrs				Th-232 $1.40 \times 10^{10}$ yrs	Th-228 1.91 yrs				Th-231 25.5 hrs		Th-227 18.7 days		
Actinium								Ac-228 6.13 hrs					Ac-227 21.8 yrs			
Radium			Ra-226 $1.62 \times 10^3$ yrs				Ra-228 5.75 yrs	Ra-224 3.66 days						Ra-223 11.4 days		
Francium																
Radon			Rn-222 3.82 days					Rn-220 55.6 sec						Rn-219 3.96 sec		
Astatine																
Polonium			Po-218 3.05 min	Po-214 $1.64 \times 10^{-4}$ sec	Po-210 138 days			Po-216 0.15 sec	64 %	Po-212 $3.0 \times 10^{-7}$ sec				Po-215 $1.78 \times 10^{-3}$ sec		
Bismuth			Bi-214 19.7 min		Bi-210 5.01 days			Bi-212 60.6 min						Bi-211 2.15 min		
Lead			Pb-214 26.8 min	Pb-210 22.3 yrs	Pb-206 stable lead (isotope)			Pb-212 10.6 hrs	36 %	Pb-208 stable lead (isotope)				Pb-211 36.1 min		Pb-207 stable lead (isotope)
Thallium										Tl-208 3.05 min						Tl-207 4.77 min

# Natural Radiation





# What is radiation exposure?

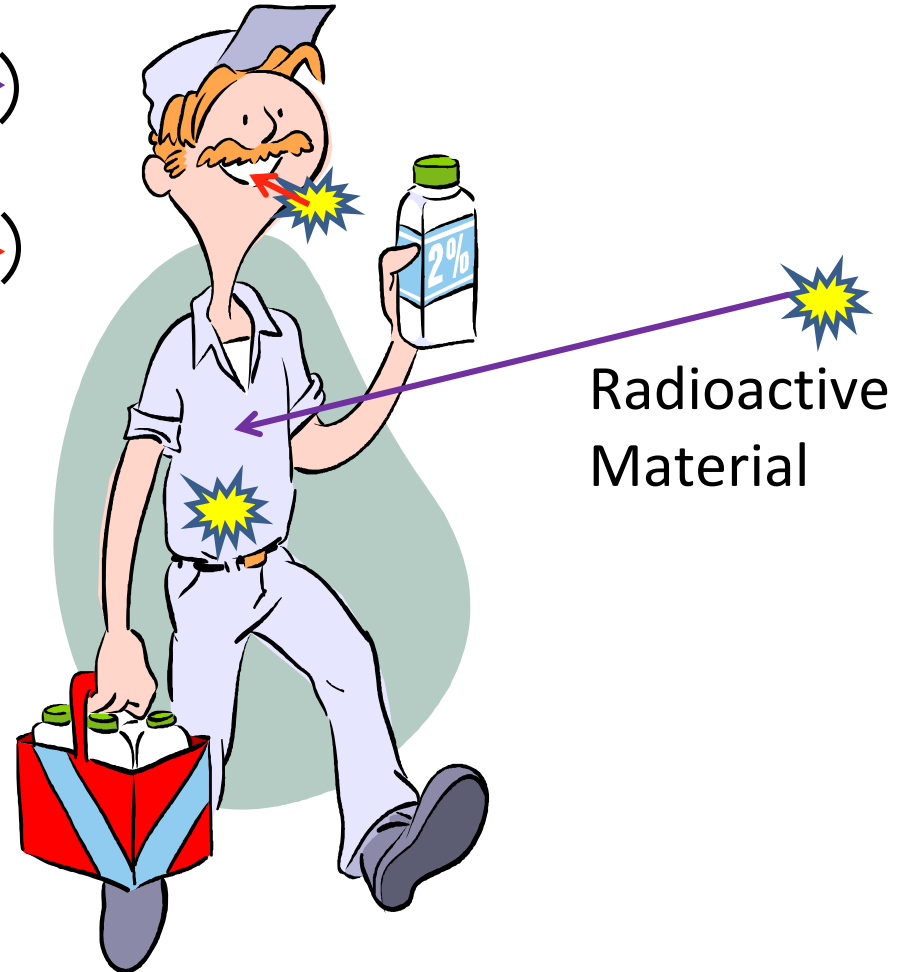
As you know now,  
You are continually  
Exposed to natural  
radiation

# Radiation exposure

- External exposure (→)
- Internal exposure (→)

**Radiation dose (Sv) :**

**External dose  
+  
Internal dose**



Sievert (Sv): A unit to measure the amount of radiation energy (J) per unit of mass (kg) considering the type and biological effects of radiation

# What is Sv, really?

Since many organs and tissues are exposed to various radiation, it is better to use “effective dose” (E), which is included the effects of radiation type and also biological effects. The unit is “Sv”.

$$\text{mSv} = 0.001 \text{ Sv}$$

# Causes of radiation exposure

External exposure + Internal exposure

Radiation dose from natural sources (mSv/y)

Mean dose value in the world:

2.4 mSv (External dose: 0.9 mSv)

Mean annual radiation dose in Japan

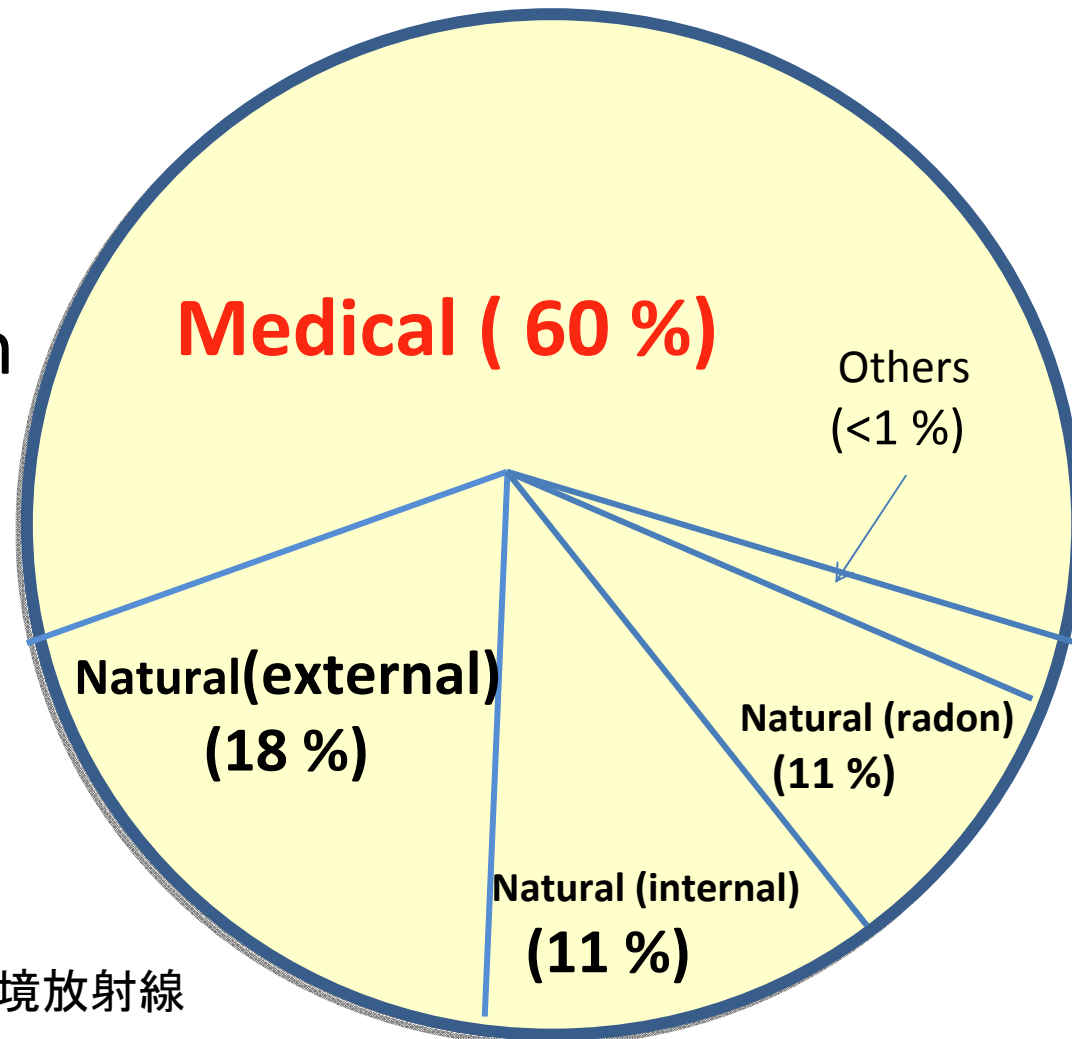
1.4 mSv (External dose: 0.7 mSv)\*

\* The value is varied to a great extent depending on the location.

# Effective doses from natural and human sources

Annual radiation dose for people in Japan

**3.75 mSv/y**



原子力安全協会(編):生活環境放射線  
(1992) をもとに作成

**When an accident happens at a  
nuclear power plant,**

**Different radioactive substances  
released from the plant are to be  
observed in the environment at various  
places depending on **distance** from the  
site and also on elapsed **time** after the  
accident.**

Typical radionuclides to be observed  
with gamma ray measurements

$^{131}\text{I}$

$^{134}, ^{137}\text{Cs}$

# Identification of the radionuclides with

Gamma energy (MeV)

Half life ( $T_{1/2}$ )

Emission ratio ( – )

$^{131}\text{I}$  ( $T_{1/2} = 8.04 \text{ d}$ )

0.080(2.6), 0.284(6.1), **0.364(81)**,

0.637(7.3), 0.723(1.8)

$^{134}\text{Cs}$  ( $T_{1/2} = 2.06 \text{ y}$ )

0.563(8.4), 0.569(15), **0.605(98)**, 0.796(85),

0.802(8.7), 1.365(3)

$^{137}\text{Cs}$  ( $T_{1/2} = 30.17 \text{ y}$ )

**0.662(90)**



# Data from the School of Engineering (since Mar. 14 2011)

1. Radiation dose rate ( $\mu$  Sv/h)
2. Detection of radionuclides in dust samples collected daily with a dust sampler

# 1. Radiation Dose Rate

Observation points:

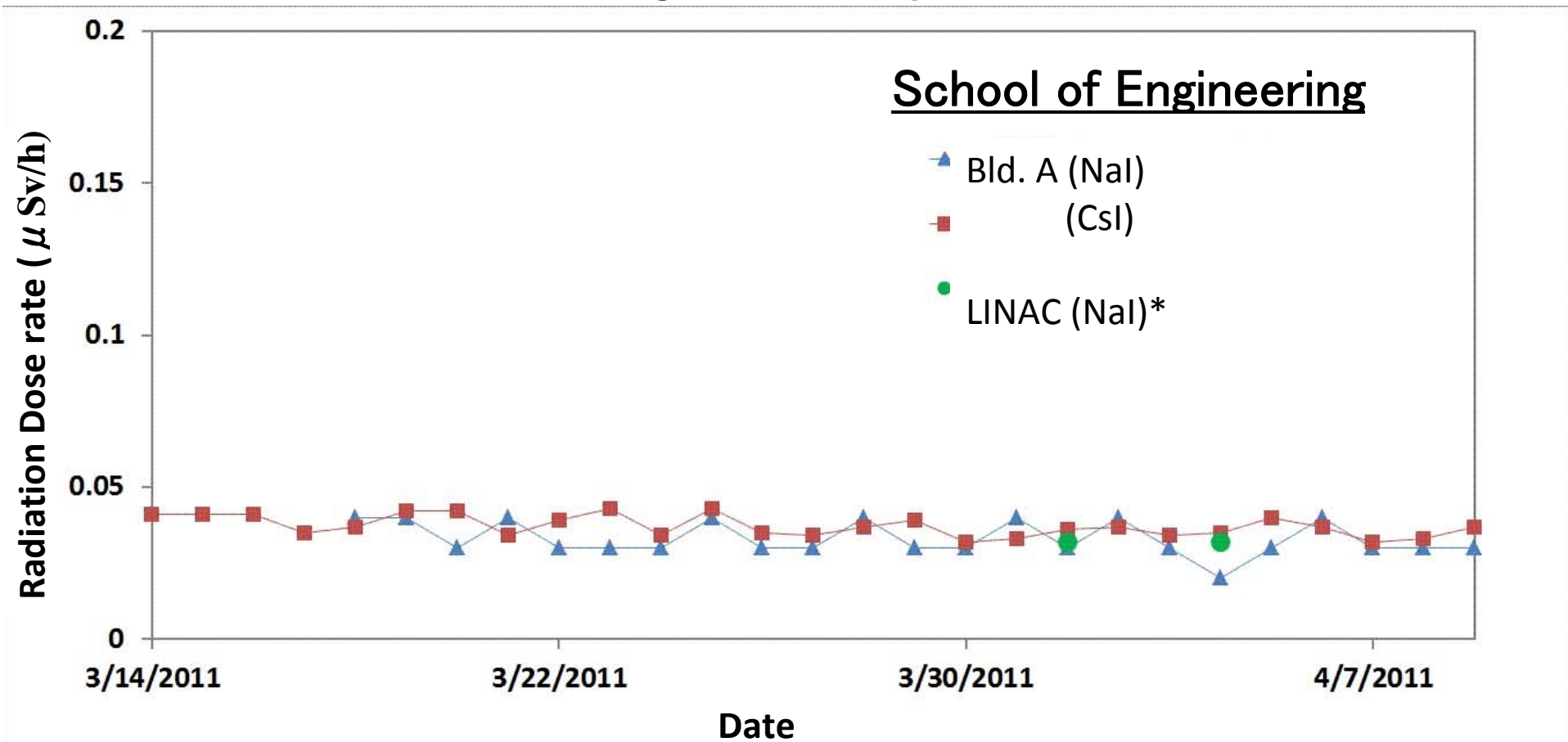
1. School of Eng. Bld. A roof  
1.3m from the floor (23.5 m  
from the ground) 1:30pm
2. School of Eng. LINAC  
Border of the controlled area  
1m from the ground 1:30pm

Instruments:

1. CsI Scintillation Counter  
(Horiba Radi PA-1000)  
NaI Scintillation Survey meter  
(Aloka, TCS-161)
2. NaI Scintillation Survey meter  
(Aloka, TCS-161)



# Time series data on radiation dose rate measured in the School of Engineering since Mar. 14 2011



\*A series of data before the accident (Mar. 11 2011) has been reserved in the LINAC facility (<http://www2.qe.eng.hokudai.ac.jp/nuclear-accident/dose/dose.html>)

## 2. Detection of Radionuclides in Dust

Observation point:

School of Eng. Bld. A roof at  
1.3m from the floor (23.5 m  
from the ground)

Dust sampling:

Dust sampler (TH-D5101/HVA-1,  
Chiyoda Technol. Co.Ltd.), Filter  
(HE-40T, ADVANTEC), 24hrs  
(Mean flow rate:700L/min)

Gamma spectrometry\*:

Dust sample (dried at 60° C for  
30min)

Gamma spectrometry (ORTEC  
GEMX10P) 20hrs



\* Conditions for the measurements (<http://www2.ge.eng.hokudai.ac.jp/nuclear-accident/dose/dose.html>)

# Radionuclides in a dust sample collected from Mar. 14 to 15 2011

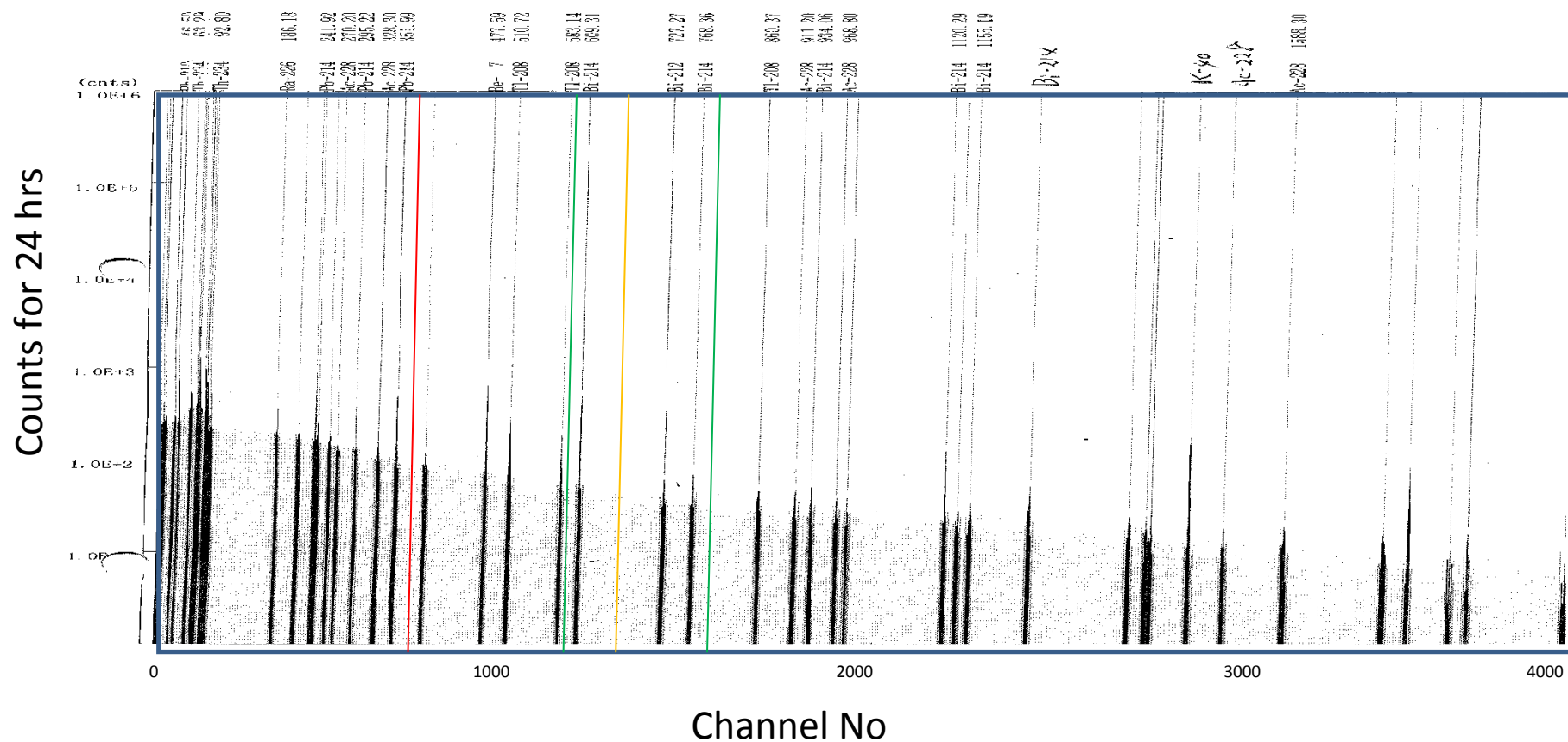
■  $^{131}\text{I}$   
■  $^{134}\text{Cs}$   
■  $^{137}\text{Cs}$

スペクトルID: Ge2  
測定日時: 2011-03-15 14-32-20

PIA No: 02  
LT=86400 sec

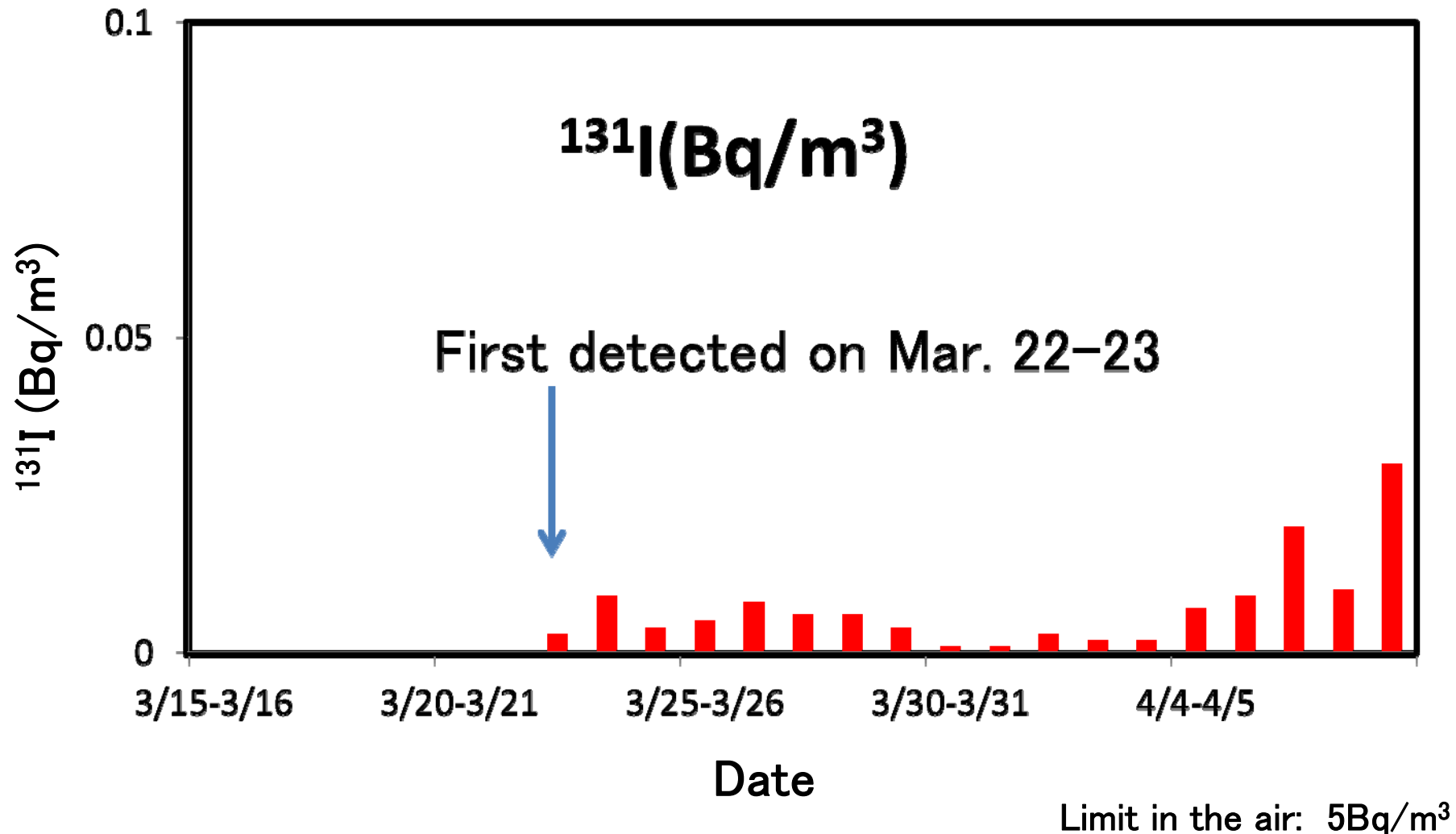
SEGMENT No: 01  
RT=86411 sec

Mar. 14-15 2011

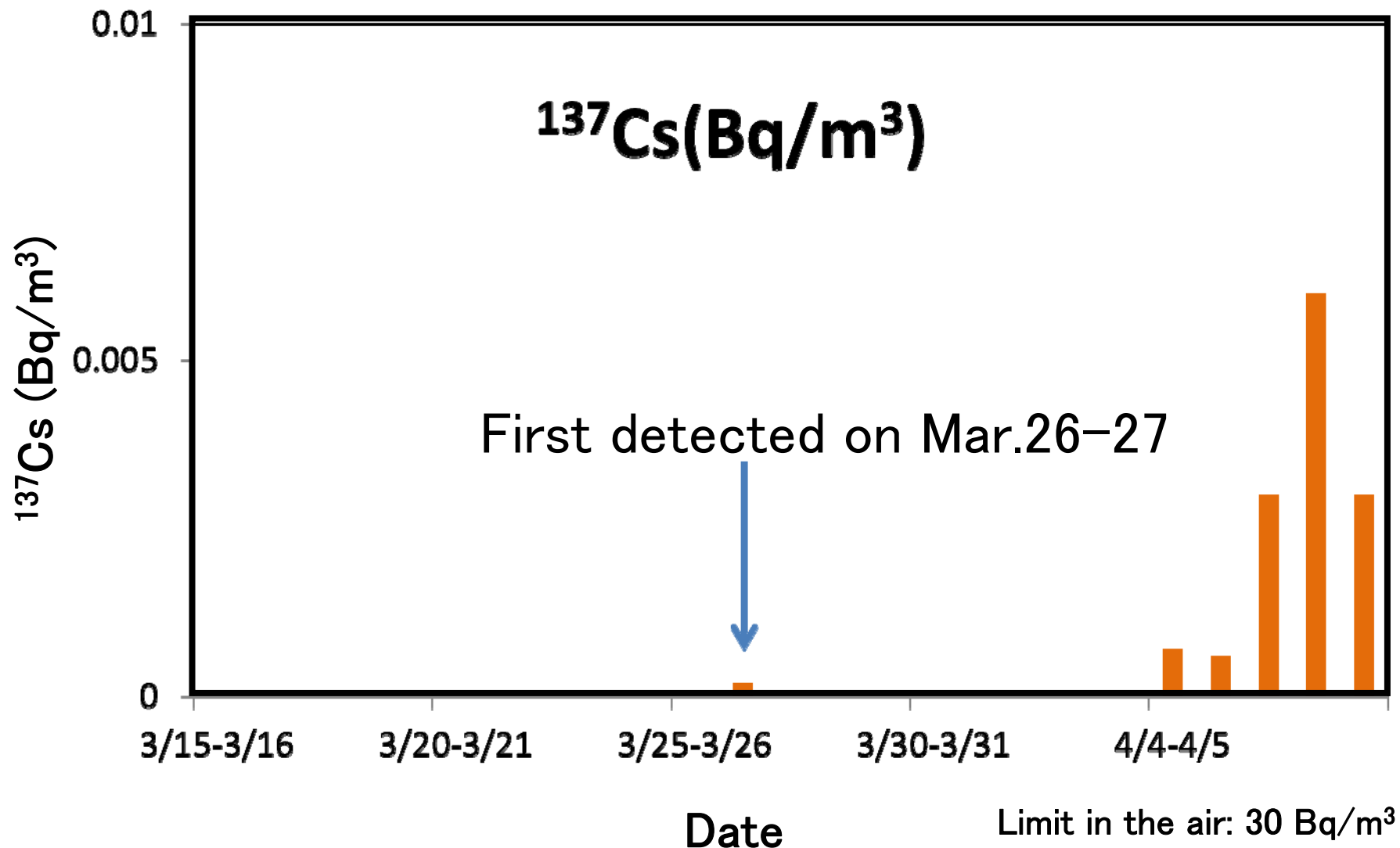




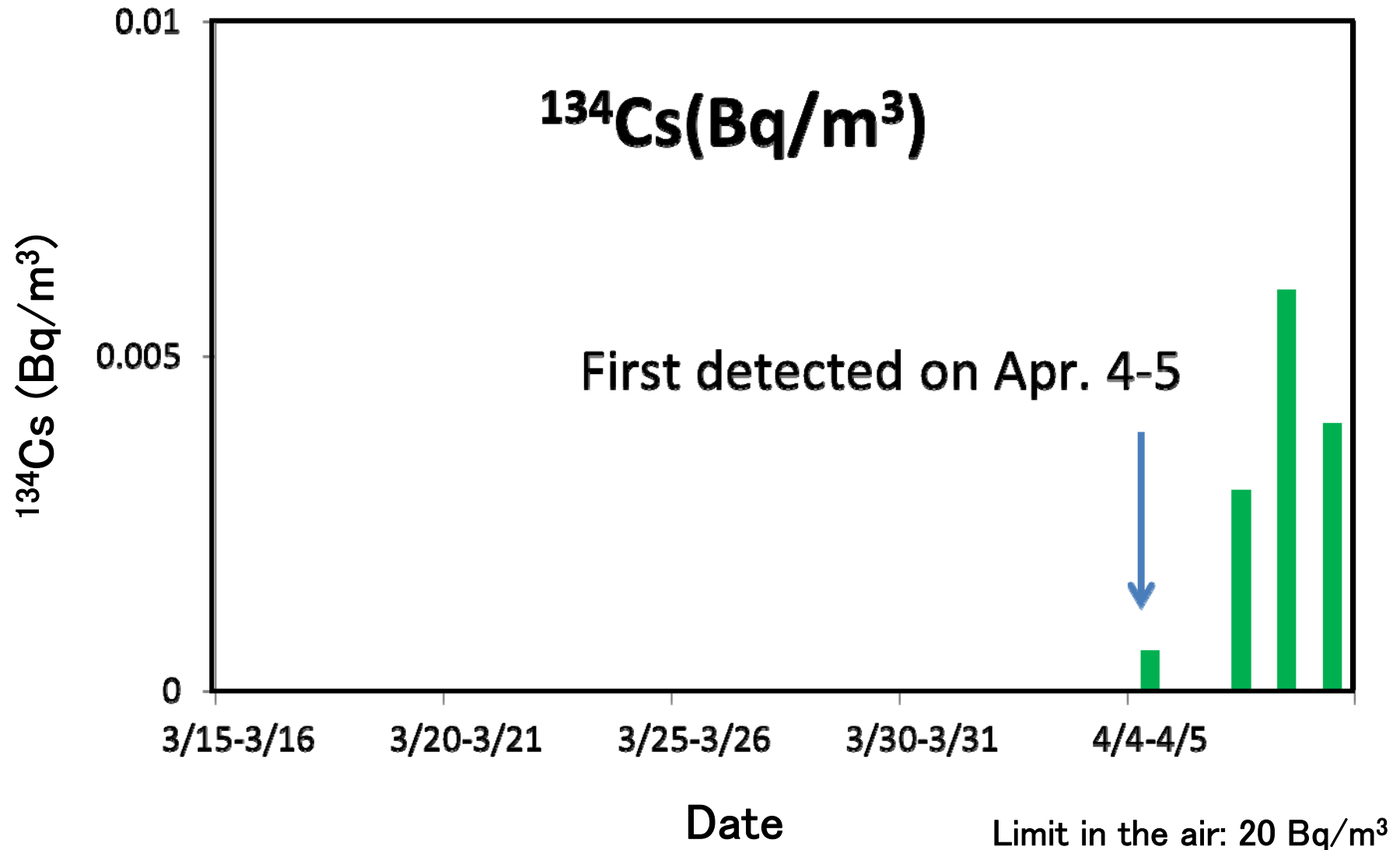
# Time series data on $^{131}\text{I}$ activity concentration in dust samples



# Time series data on $^{137}\text{Cs}$ activity concentration in dust samples



# Time series data on $^{134}\text{Cs}$ activity concentration in dust samples

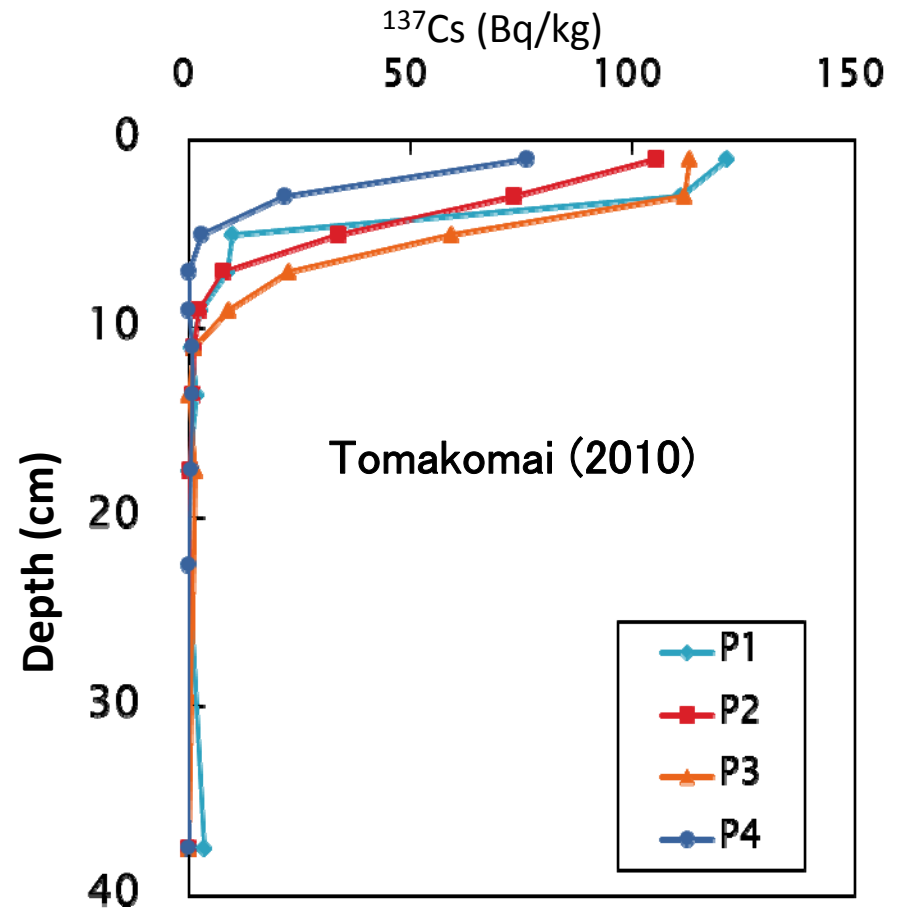
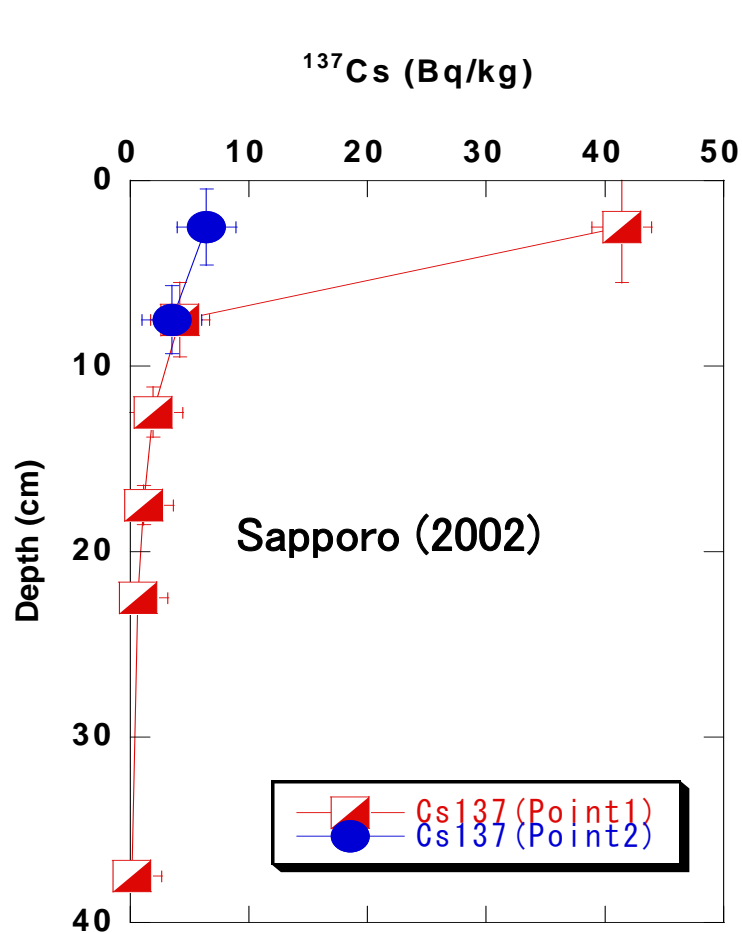




## Data before the accident

1. Radiation dose measured at LINAC facility since 2006
2. Activity concentration of  $^{137}\text{Cs}$  in soil on the campus of Hokkaido University in 2002

# Depth profiles of $^{137}\text{Cs}$ activity concentrations in soil



## Findings:

1. Radiation dose rate:

0.03–0.04  $\mu$  Sv/h

(Mar. 14–Apr. 10)

2. Detection of radionuclides in dust

Mar. 22–23:  $^{131}\text{I}$

Mar. 26–27:  $^{131}\text{I}$ ,  $^{137}\text{Cs}$

Apr. 4–5 :  $^{131}\text{I}$ ,  $^{137}\text{Cs}$ ,  $^{134}\text{Cs}$

# Summary

- There are various radioactive substances in the environment.
- The unit of radioactivity: Bq
- The unit of radiation dose: Sv
- Data from the School of Engineering  
Radiation dose rate: 0.03–0.04  $\mu$  Sv/h  
Radionuclides detected:  $^{131}\text{I}$ ,  $^{134}\text{Cs}$ ,  $^{137}\text{Cs}$   
(The activity level is quite low)
- Monitoring has still been continued.

# Examples of useful HP:

<http://www2.ge.eng.hokudai.ac.jp/nuclear-accident/dose/dose.html> (Data)

<http://www.iaea.org/news/tsunamiupdate01.html> (Daily report)

[http://ec.europa.eu/energy/nuclear/radiation\\_protection/publications\\_en.htm](http://ec.europa.eu/energy/nuclear/radiation_protection/publications_en.htm)  
(Basic knowledge on radiation)

<http://www.nirs.go.jp/ENG/index.html> (Radiation protection)

and more...

# Thank you for your attention!

Ryoko Fujiyoshi